| DATE | PERIOD |
|--|--|
| Cell Membrane & Tonicity Worksheet | |
| Composition of the Cell Membrane & Functions The cell membrane is also called the | SKETCH AND LABEL a phospholipid coloring the heads red and the tails blue. |
| Another type of lipid in the cell membrane is that makes the membrane membrane membrane is that also aid in diffusion and in cell reference in the phospholipid bilayer are that also aid in diffusion and in cell reference in the proteins go all the way through the bilayer, while only on one side. Integral proteins are also called proteins. Large molecules like carbohydrates use proteins to help move across cell membranes. Some of the membrane per carbohydrate attached to help cells in recognize each other and certain molecules that it is a functions of the cell or plasma membrane: a b d. | proteins are e or roteins have |
| Correctly color code and identify the name for each part of the cell membrane. Letter Name/Color Letter Name/Color Phospholipid bilayer (no color) Peripheral protein (red) Integral protein (pink) Cholesterol (blue) Fatty acid tails (orange) Glycoprotein (green) Phosphate heads (yellow) Glycolipids (purple) | |
| | nbrane diagram. re/Function |
| Attracts water Repels water Helps maintain flexibility of membrane Make up the bilayer Involved in cell-to-cell recognition Help transport certa cell membrane | in materials across the |

NAME___

| Define osmosis. | | | | |
|---|---|---|--|------------------------------|
| In which direction does water move across n | nembranes, up o | or down the cor | ncentration gradien | it? |
| Define these 3 terms: | | | ********* | |
| a isotonic- | | | | |
| b. hypertonic | | | | |
| c. hypotonic | | | | |
| Use arrows to show the direction of water m isotonic environment light blue, the hypoton green. | ovement into or ic environment | r out of each ce yellow, and the | ell. Color and label to hypertonic enviror | he cell in an nment light |
| 98% WATER | 98% WATER | | MATER WATER | |
| Match the description or picture with the osn A. Isotonic | solution w | | ate concentration | |
| | condition ¡ | olant cells requ | | the same |
| C. Hypotonic | plant cell lo solution w plant cell v | cell bursts (cyto oses turgor pre ith a higher sol vith good turgo | olysis) ssure (Plasmolysis ute concentration |) |
| Label the tonicity for each solution (isotonic, | | - | | |
| и, о н, о н, о | H ₂ O | | H,Q | H ₂ O |
| Transport Requiring Energy What type of transport is represented by the | | | OUTSIDE CELL | |
| following picture? | | | 0 | |
| What energy is being used? | the | (() | | \ |
| Color the internal environment of the cell yello Color and Label the transport proteins red and substance being moved blue. | | | INSIDE | |

| One type of active transport is called the pump which helps muscle cells contract. This pump uses to move ions the concentration gradient. The protein that is used to pump the ions through is called a protein and it changes its to move the ions across the cell membrane. Label and color the carrier proteins red and the ions green. | apssure (xx) | day of the state o | (A) |
|--|--------------------------------------|--|---|
| TONICITY AND OSMOSIS | | | • |
| key: solute particle * cell inembrane cell well | B | c | plant cell |
| Part I ~ Fill in the blanks. A is a fluid in which a substance dissolve A is a combination of sol The process by which water diffuses across a membra | d in a solvent. ute and solvent. | | |
| Part II – Look at the solutions illustrated above and | fill in the blanks. | | |
| 1. Solution B is to Solutio | n A. This is because S | olution B has a gre | ater |
| concentration of in it than do | es Solution A. Solu tion | on C has no solutes | dissolved in it, |
| therefore it is to both Solu | tions A and B. | | |
| 2. As a relative concentration of solutes in two solutions Solution does Solution C; Solution A is also hypertonic to Solution. | n A has a lower conce | - | |
| 3. If you wanted to make Solution A isotonic to Solu | tion B, you could add | water to Solution | or you |
| could add solute to Solution . If you took all thre | | | |
| them thoroughly, then redistributed the solution am | | _ | |
| Solution B. Solution A would also be | | | |
| Solution B. | | | |
| Part III – Look at the solutions and cells illustrated a | have and fill in the h | lanke | |
| Because the cytoplasms of the plant and the animal street in the cytoplasms. | | | itas wasan say |
| their cytoplasms are to each other It | | | |

Solution A, we would expect no change in the cells, because Solution A is to the cytoplasm of each cell. 2. Let's put both cells into Solution B. Because Solution B is hypertonic to the cytoplasms of the cells, we would expect water to ______ the cells through the process of ______. This would result in the cytoplasm of both cells shrinking. 3. Now we'll put both the plant and animal cell into Solution C, which, because it contains no solutes at all, is to the cytoplasm of both cells. will enter both cells through osmosis. The animal cell is likely to ______, unfortunately. The plant cell, however, is protected from this because of the presence of its _____ **Cell in Hypertonic Solution** Cell in Isotonic Solution Cell in Hypotonic Solution After osmosis Before osmosis ✓ ○ ○ Water molecules Dissolved particles Water molecule Sugar molecule Selectively permeable , membrane

Refer to the U-tube pictures above when answering the questions below.

- 1. Why did the number of water molecules on each side of the membrane change, whereas the number of sugar molecules stayed the same?
- 2. How does the plasma membrane of a cell compare with the membrane in the U-shaped tube?
- 3. Explain the behavior of water molecules in the isotonic solution.
- 4. Does osmosis occur if a cell is placed in an isotonic solution?
- 5. Why does water enter a cell that is placed in a hypotonic solution?
- 6. What happens to the pressure inside a cell that is placed in a hypertonic solution?
- 7. What can happen to animal cells when placed in a hypotonic solution? Explain.
- 8. What causes a plant to wilt?